Les Mardis de la Chimie Durable

ARKEMA



Présentation de travaux en chimie durable Spéciale « Industrie* » 4^{ème} édition en wébinaire















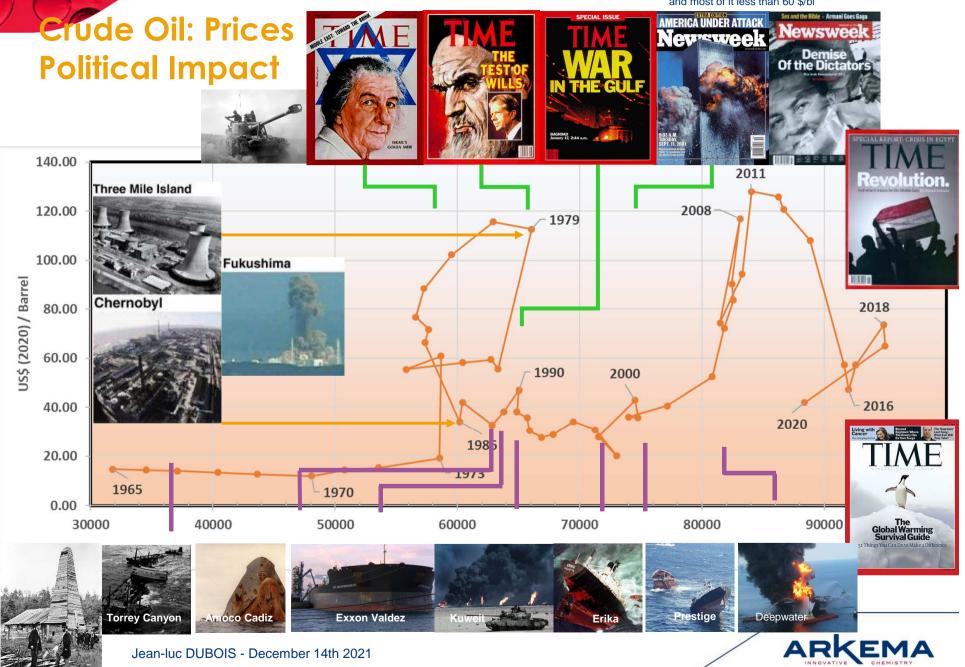




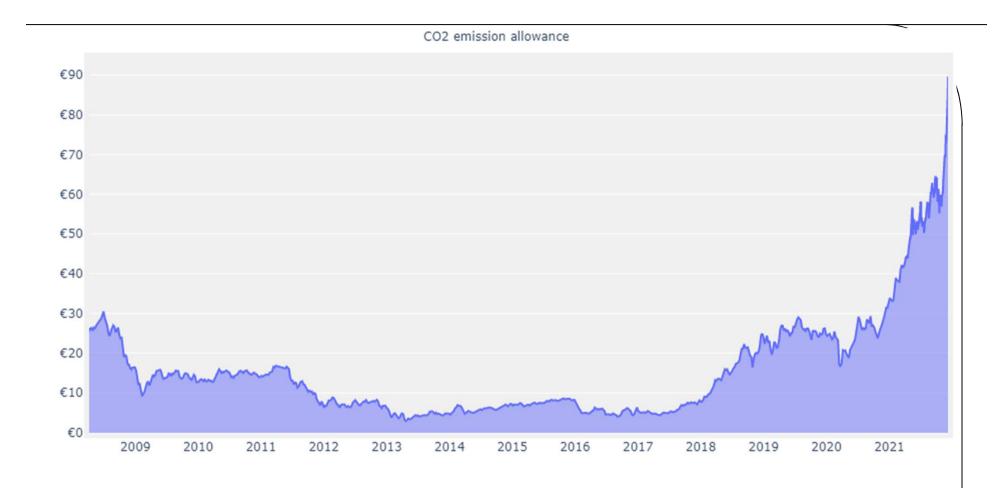
Chimie Durable: conversion de biomasse, recyclage et conversion du CO2. Exemples et challenges

Jean-Luc Dubois (Directeur Scientifique, Arkema)

Crude oil production costs less than 30\$/bl in middle east, and most of it less than 60 \$/bl



CO2 emission allowance €/tonne

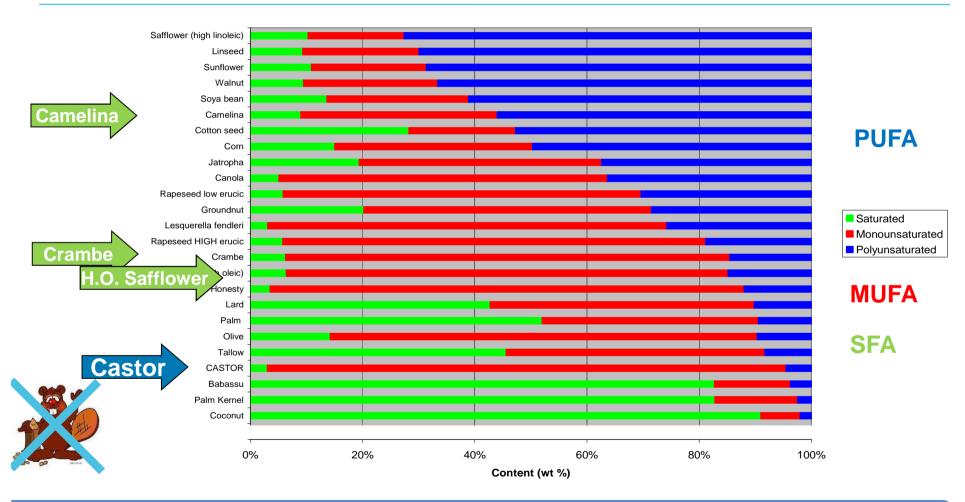


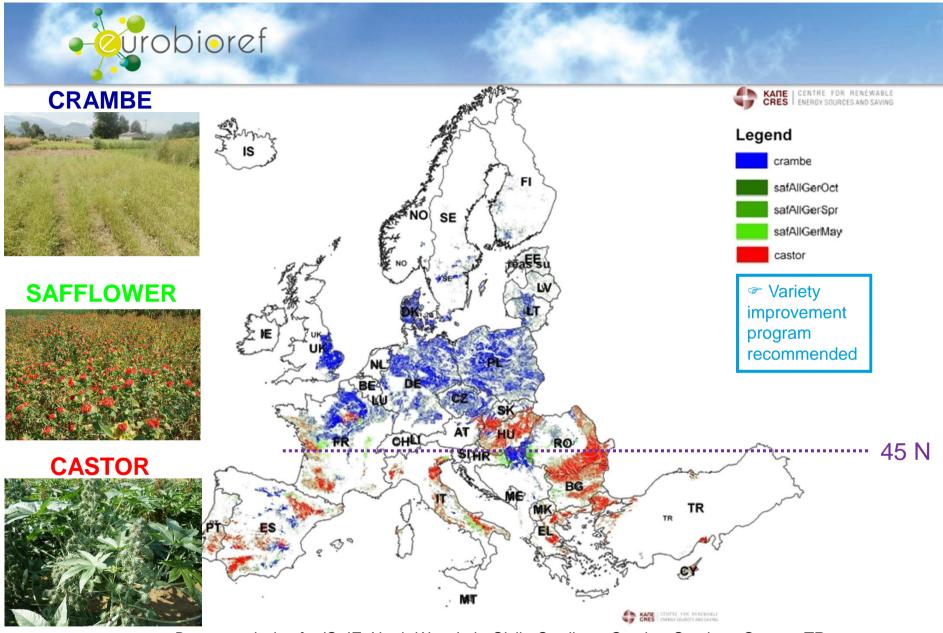
→ 100€/t CO2 → about 50 \$/bl more on crude oil price



Fatty Acid Profile

Source: « Lexicon of Lipid Nutrition », Pure Appl. Chem., Vol 73, N°4, PP 685-744, 2001 and Oleon datasheets + internal data





Data are missing for IS, IE, North-West Italy, Sicily, Sardigna, Corsica, Southern Greece, TR Contribution from Myrsini Christou and Giannis Eleftheriadis (CRES)



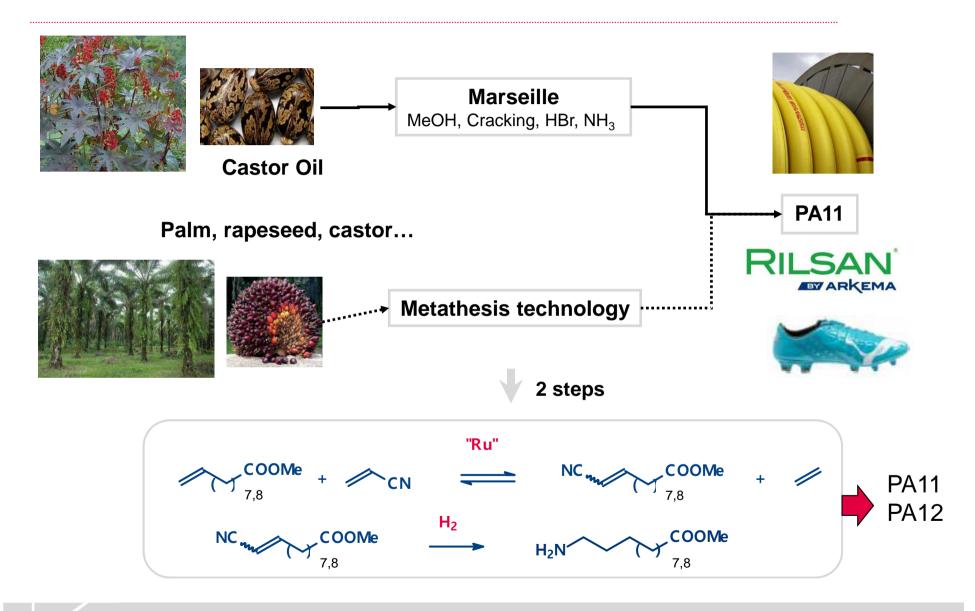
Metathesis of Vegetable Oils for Aminoacids production







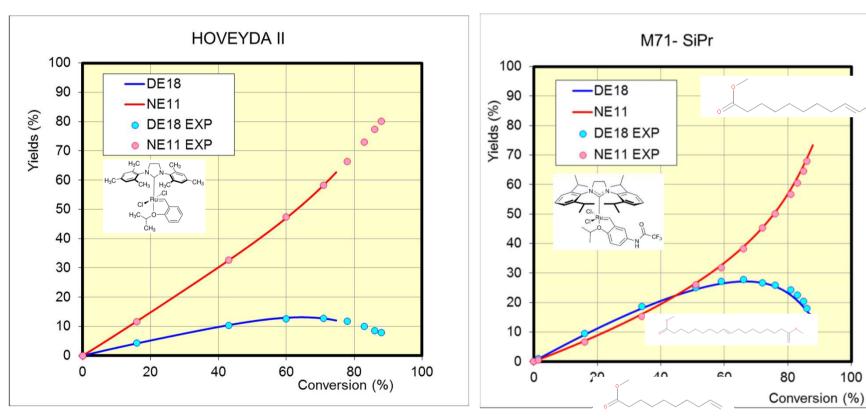
ALTERNATIVE MONOMERS ROUTE



KINETIC MODELIZATION, SIPR VERSUS SIMES



Conditions: continuous catalyst addition (100 and 30 ppm), half acrylonitrile at start of reaction and continuous addition, Toluene solvent, reflux condition, 10 wt % 9-methyl decenoate in Toluene



In general, SiPr catalysts are more active and SiMes Catalysts are more selective



100

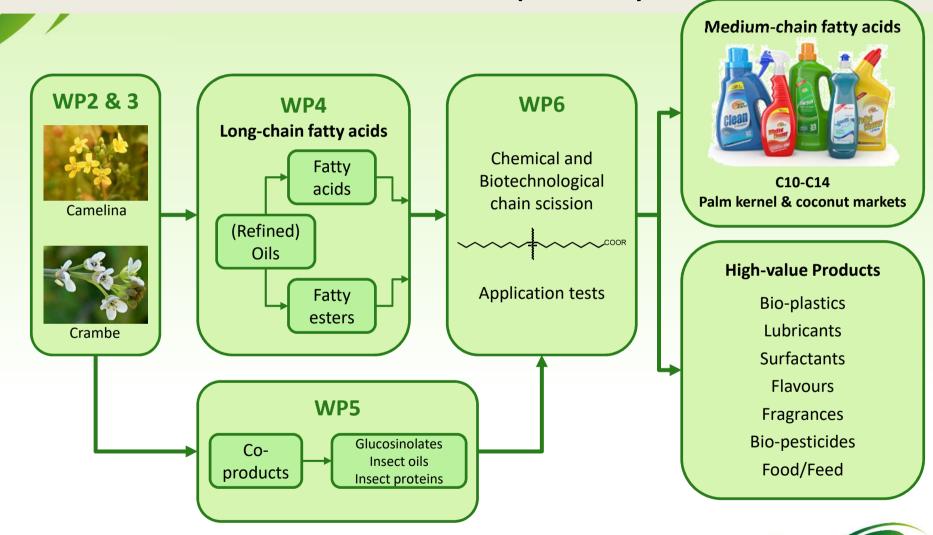


Castor based PA12 sample Made with Arkema's Methyl-10-undecenoate.



Jean-luc DUBOIS - December 14th 2021

COSMOS: Camelina and crambe Oil crops as Sources for Medium-chain Oils for Specialty oleochemicals



Vegetable Oils Fatty Acids Profiles

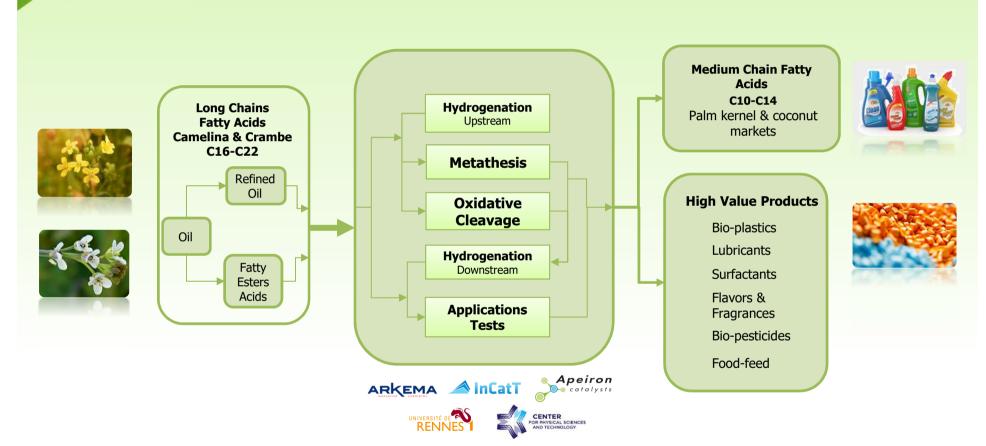
CX:n (X=chain length, n=number of unsaturation or C=C bonds)

Oil	C8:0	C10:0	C12:0	C14:0	C16:0	C18:0	C18:1	C18:2	C18:3	C20:0	C20:1	C22:0	C22:1
Rapeseed					4.5	2	57.5	23	11	0.6	2.3		
Soybean					9.0	4.5	25	49.5	11				
Palm Kernel	3.5	3.0	50.5	15.5	9.0	2.0	14.5	2.5		0.5			
Coconut	8	7	48	16	9	2	7	2					
Camelina					6	2	15	18	35	1	19		1
HEAR					2.5	2	17	18			10	1	45
Crambe					2.4	0.8	17.9	8.3	5.7	0.5	4.2	2.1	55.9

Need solutions to make short fatty acids from European crops



Catalysis in Cosmos





Fatty Acids conversions and applications





Oxidative cleavage of MUFA:

Cross-metathesis (ethenolysis, butenolysis, acrylonitrile):

COOR
$$\stackrel{R}{\longrightarrow}$$
 $\stackrel{R}{\longrightarrow}$ $\stackrel{R}{\longrightarrow}$



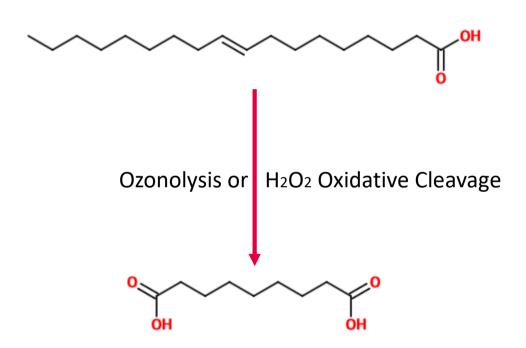
9-CARBONS DIACID: AZELAIC ACID

CRODA SIPO

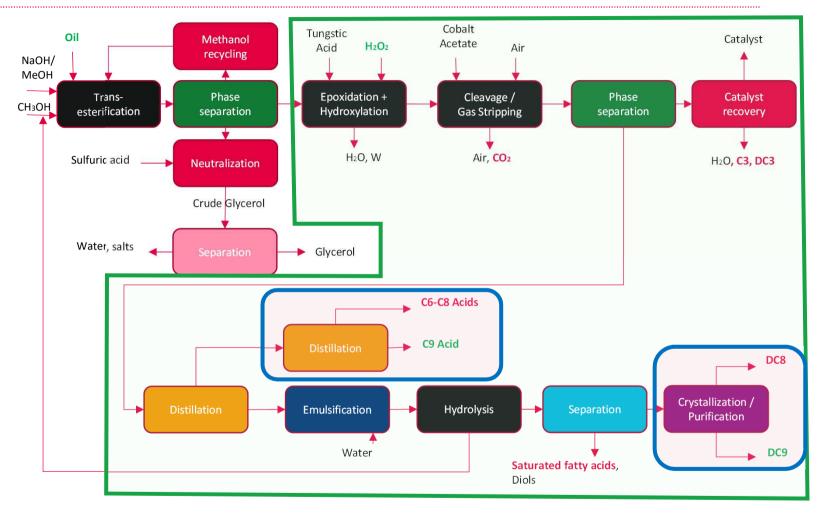




MATRICA



PROCESS BLOCK DIAGRAM: 3 SCENARIOS FROM THE OIL, FROM THE ESTER, WITHOUT FINAL PURIFICATIONS





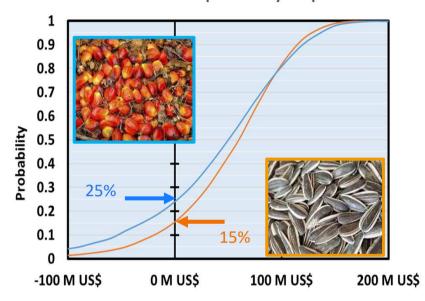




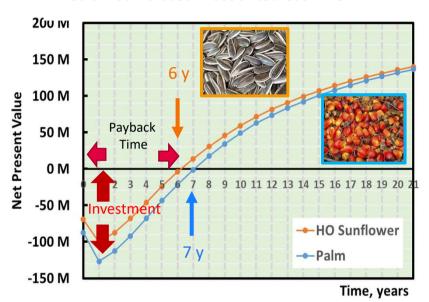


NET PRESENT VALUE, PAYBACK TIME, RISK ANALYSIS

Net Present Value probability frequencies



Median Cumulated Discounted Cash Flow



Net Present Value after 10 years of production

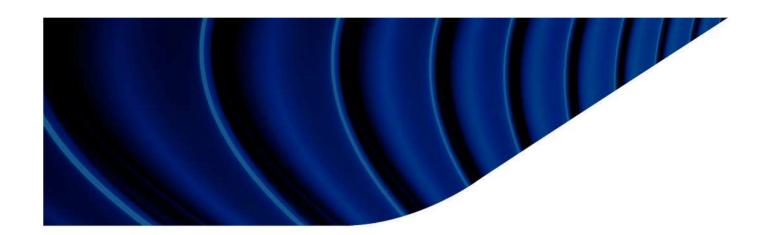
- Net Present Value: Probability to make more money than a reference case at 10 % internal return.
- Payback time: time needed to cover Investment costs
- High Oleic Sunflower Oil, although taken at higher price, is a less risky option. But still requires much time for Payback (Risk on Capital).









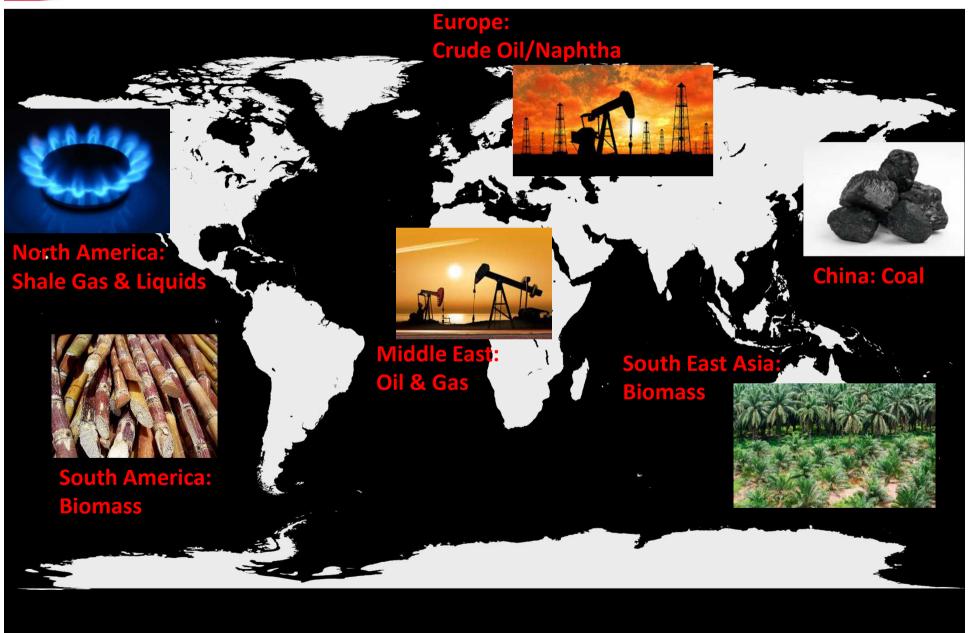


Alternative routes to Acrylic Acid

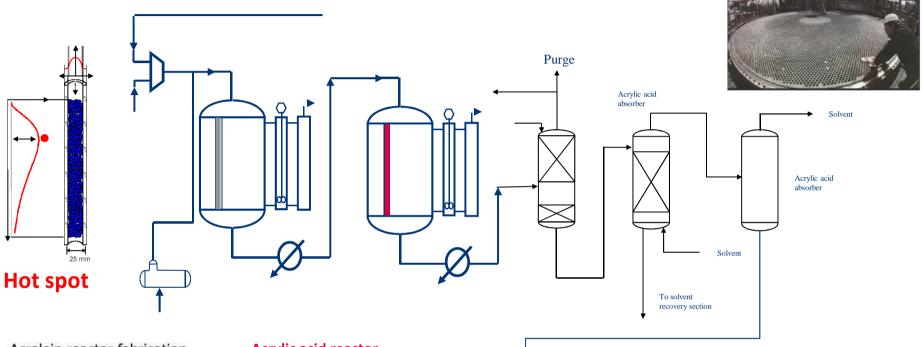




Process selection may depend on local feedstock availability.



Conventionnal Propylene oxidation process
Multitubular Fixed Bed Reactors



Acrolein reactor fabrication



Number of tubes: > 27.000

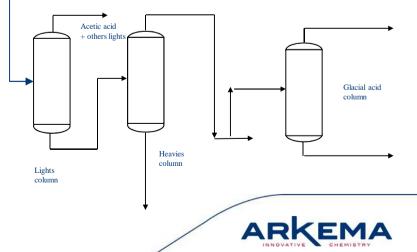
Weight: 350 MT Multitubular fixed bed reactors @ dwe.com

Acrylic acid reactor



Number of tubes > 27.000 Weight: 570 MT

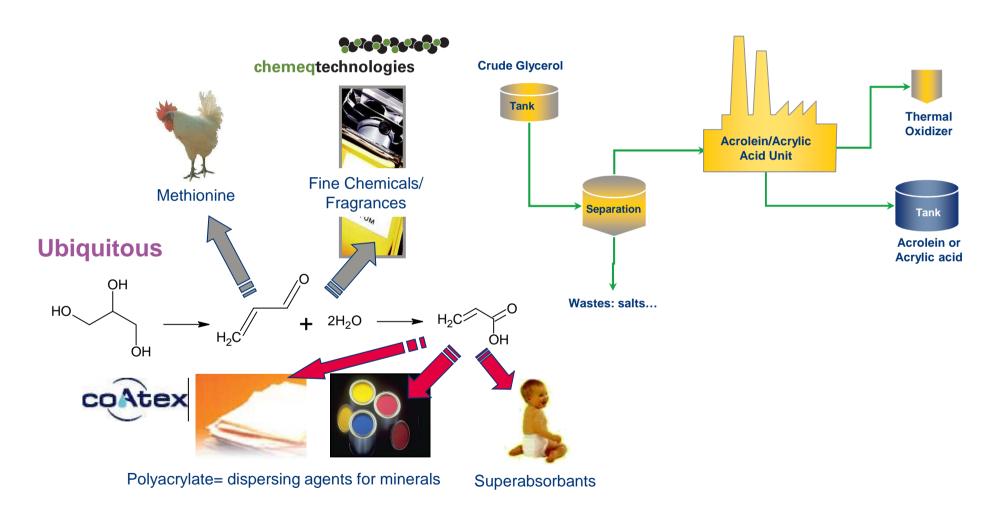
Jean-luc DUBOIS - December 14th 2021





Acrolein/ Acrylic Acid: ARKEMA's Project

┏ Double internal dehydration of Glycerol leads to Acrolein, further oxidized to Acrylic acid





Environmentally Friendly Process: Avoid scenarios of past accidents

Ardeche

 Target: develop a new process for on-site Acrolein production to avoid storage and transportation of a highly toxic Chemical.

Drome

- Pierre Bénite, France. July 10, 1976.
- Taft, USA/ December 10, 1982.



Major contamination of the Rhône River, during clean-up of a train tank 367 Tons dead fish The Daily Record, Ellensburg, Wash., Sat., Dec. 11, 1982 Page 5

Explosion rocks Louisiana plant

TAFT, La. (UPI) — A fiery explosion of a storage tank at a chemical plant today shook south Louisiana's industrial corridor, forcing the evacuation of up to 25,000 people and the closing of the least Mississippi River to ship traffic.

No injuries or deaths were reported in the explosion and resulting fire at the Union Carbide plant. Company officials said the storage tank was surrounded by protective mounds of dirt.

"The immediate concern is the adjacent tanks that have additional explosion potential," Union Carbide spokesman Jim Tate said from a concrete command post on the plant grounds.

"As a precautionary measure and to the minimize the risk to the surrounding areas, an evacuation has been conducted by the state police." Civil Defense officials said as many as 25,000 people had been forced from their homes.

The Coast Guard said it closed the river to ship traffic, describing the move as a precaution against Acrolein fumes, although no immediate danger was reported. The chemical is a yellowish or colorless pungent liquid sometimes used for the production of tear gas.

17,000 Louisiana Residents Return Home as Site of Blast Is Declared Safe Aerial view of the wreckage of tanks after an explosion and fire Saturday at the Union Carbide chemical plant in Taft, La. Residents within a five-mile radius of the site were evacuated because of the fear of toxic fumes from the fire. They were permitted to return home yes-terday. River traffic near the plant, situated on the banks of the Mississippi River 30 miles west of New Orleans, was halted until fire burned itself out. The chemical in the tanks was acrolein, which is used to make algaecides, animal food supplements and tear gas.

The New Hork Times

Explosion of a Storage facility 17000 people evacuated





Alternative Routes to Acrolein & Acrylic Acid

ARKEMA Mixed alcohol direct oxidation via acrolein (2010-)

Cross metathesis (ethenolysis) of Fumaric or Muconic acids (2000s)

Carbonylation of Ethylene Oxide via Propiolactone (mid 2000s-)

Via Fermentation to 3-Hydroxypropionic acid /PHP (mid 2000s-)

Gas to olefins (via syngas and CH₃OH)

ARKEMA Glycerol oxydehydration via acrolein (early 2000s)

Methanol to olefins/propylene

ARKEMA Propane direct oxidation (1990-early 2000s)

Propane dehydrogenation to propylene

ARKEMA Propane – Ballast/Inert gas (1987-)

Propylene oxidation (2 steps), 1970s

Lactic acid dehydration (1950s-)

Propylene Oxidation (1 step)

Acrylonitrile Hydrolysis

Acetic acid to ketene+Formaldehyde, to propiolactone, 1960s (not commercialized)

High T, High P, Reppe process, 1956 (Nickel free)

Reppe Chemistry, Acetylen+CO,+ROH 1954

Ethylene Cyanohydrin process, 1927



ARKEMA











OPPORTUNITIES AND TECHNICAL CHALLENGES

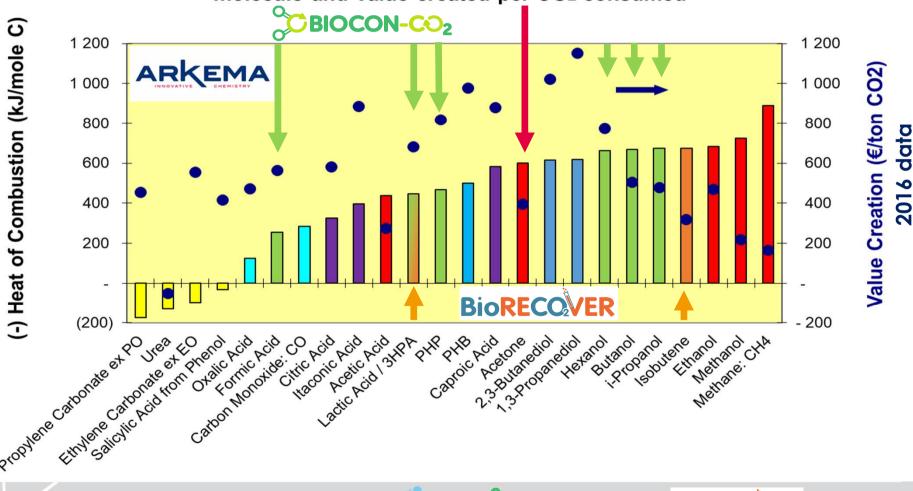




TARGET PRODUCT SELECTION: VALUE CREATION AND ENERGY CONSUMPTION

PYRCCO₂

Heat of Combustion as image of energy consumed to produce the molecule and Value created per CO2 consumed

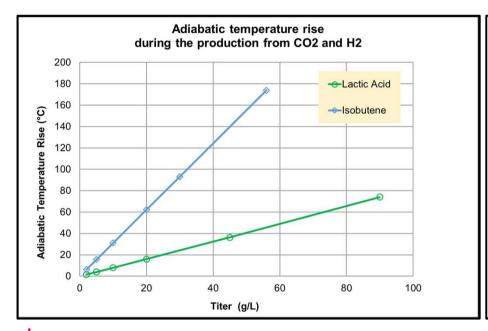


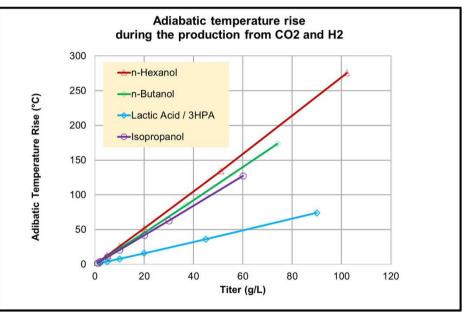






HEAT MANAGEMENT: ADIABATIC TEMPERATURE RISE





- Probably impossible to reach more than 10-20 g/L without external cooling
- Low Titers = High Capital Cost
- 100 kt/year of lactic acid = energy loss of 93 900 MWh/year = equivalent of the energy consumed by more than 14 000 Europeans in their households.











This project has received funding from the European Union's Horizon 2020 research and innovation program under grant agreement N° 820687.

Price (€/kg) High Medium High Price (€/kg) Price (€/kg

1000 kt/y

Polymers European market shares, Plastics Europe Plastics converter demand by resin type (2016)

Source: JL DUBOIS -November 20th 2018

100 kt/y

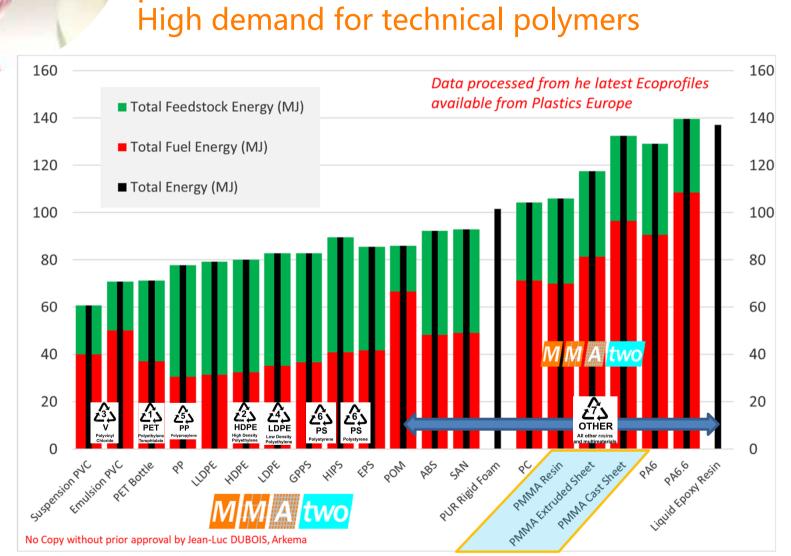


Medium Low

Low

10 000 kt/y

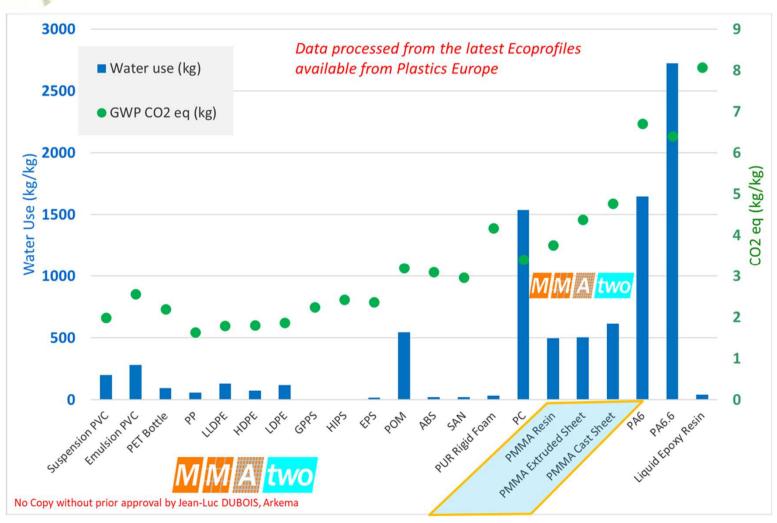
Energy Consumption in Polymers production:





\ e

Water consumption and CO₂ eq emissions







350-400°F

-123

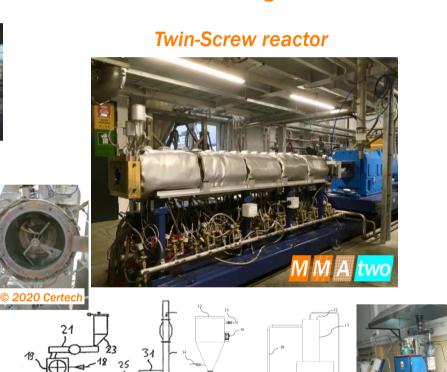




Rotating Drum reactor

Microwave & Induction

reactors



MMA <mark>two</mark>

Auger reactor

(w or w/o circulating solid)

Fluid Bed reactor

Fig.1 Apparatus

Picture borrowe

Condenser Pyrolyzer Hot

From Recycling Tecl

ARKEMA INNOVATIVE CHEMISTRY

Molten Metal reactor

Hot sand

Hot air

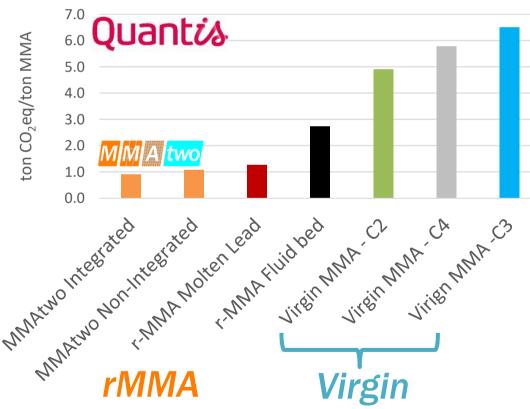
Regenerator

Rotating Paddle reactor



LCA: M36 Main results

Benchmarking with virgin technologies



Main Results

- Compared to virgin production, there is a large impact reduction (more than 75%) depending on the technology
- technologies, the impact reduction ranges between -20% to -60%
- More recycling processes to come soon

Data sets:

Virgin MMA: C2 route (3), C3 route (3), C4 route (3) R-MMA: Lead Bath (3+1); Fluid Bed (2), Rotating Drum (3), Dry Distillation (5), Stirred Tank (3) Dry Distillation with biomass as energy Source (3)







Chain of Custody:

Chain of Custody: Options



Customer demand

01

Identity Preserved, Dedicated plant

Regenerated MMA 100 %



02

Segregated, Single plant Batchwise Regenerated MMA 100 %

Virgin MMA

O3 Cor

Controlled Blending, Mixed stream, single product

Regenerated MMA 50%

04

Mass Balance, Mixed streams. Allocation, Physical link Regenerated MMA 25%

Regenerated MMA 75 %

05

Book & Claim Credits and Products follow different path



« Green Notes for Green Chemistry »



- Development of « Green » chemistry requires to set the conditions to make money
- Use all parts of the crops
- Develop Safer processes
- Renewable chemicals with improved Life Cycle Analysis
- Renewable (bio-based) ≠ biodegradable ≠ non-toxic
- High value polymers offer more opportunities for recycling

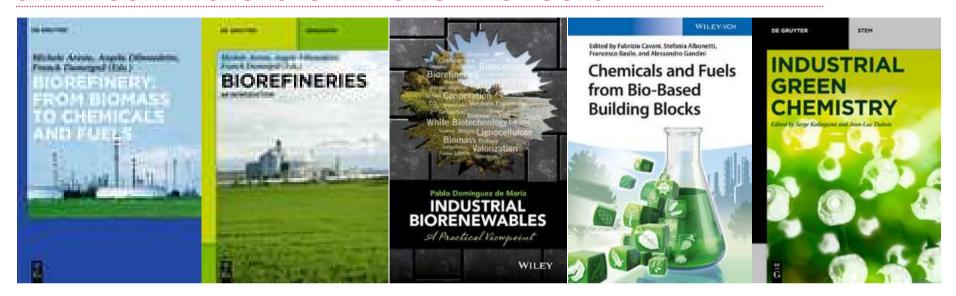






TO LEARN MORE:

SEE MY CONTRIBUTIONS TO THE FOLLOWING BOOKS



Chapter 2: Refinery of the future: feedstock, processes, products (2012) Chapter 10: Oil chemistry: chemicals, polymers, and fuels (2015) Chapter: Castor Reactive Seed Crushing Process to Promote Castor Cultivation (2016) Chapter: Arkema's Integrated Plant-based Factories (2016) Chapter 2
Alternative routes to more sustainable acrylonitrile: biosourced acrylonitrile.
Chapter 4 Fatty nitrile esters hydrogenation for biosourced polyamide polymers.

Thank you for your attention

