Light-induced CO₂ reduction catalysis with urea-modified iron porphyrin

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Finding ways to valorize and transform CO_2 into fuel using renewable energies as an alternative to fossil fuels is crucial in the current scientific research. A possible approach is electro or light induced molecular catalysis. Iron porphyrins had been reported to be active catalysts for CO_2 electroreduction since the $80s^{1}$. Our group has previously developed a highly active iron porphyrin catalyst bearing urea groups in the second coordination sphere $(UrFe)^2$. The use of the urea scaffold to stabilize the CO_2 adduct enabled the catalyst to display low overpotentials and high turnover frequency. These results lead us to investigate its catalytic activity in a light induced catalytic system. When using ruthenium tris-bipyridine as a photosensitizer, our study shows that the second coordination sphere effect can also be transposed to homogeneous photocatalysis for CO_2 reduction. Careful tuning of the photocatalytic reaction parameters and modification of second coordination sphere led to a great enhancement of catalyst durability, with one of the highest turnover numbers (TON>7000) reported in the literature, and excellent selectivity for CO (>99%)³.



Figure 1. Simplified scheme of CO_2 -to-CO photocatalytic reduction using $[Ru(bpy)_3]^{2+}$ as photosensitiser and FeUr as catalyst.

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